

# DENSITY FUNCTIONAL THEORY-BASED STUDY OF $\text{Cu}_2\text{TiSnS}_4$ AND $\text{Cu}_2\text{VSnS}_4$ FOR PHOTOVOLTAIC APPLICATIONS

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## I. INTRODUCTION

$\text{Cu}_2\text{ZnSnS}_4$  (CZTS) is a promising earth abundant and non-toxic intrinsic p-type semiconducting material [1]. However, similar radii of Cu and Zn atoms facilitates formation of anti-site defects which degrade performance of CZTS solar cells [2]. Thus, there is need to find an alternative to Zn to reduce cation disordering. In the present work, we fully replace Zn atoms in CZTS by Ti and V atoms and study the effects of the substitution on the band structure and stability of the alternative kesterite material using density function theory (DFT) [3].

## II. METHODS AND RESULTS

The calculations have been performed by Vienna Ab initio Simulation Package (VASP). It has been found that upon Zn substitution by Ti in CZTS its band gap ( $E_g$ ) decreases and shifts from direct to indirect compared to that of pristine CZTS (Fig. 1a-b). On the other hand, there is no  $E_g$  upon Zn substitution by V (Fig. 1c). The calculated binding energies of CZTS (10.16 eV),  $\text{Cu}_2\text{TiSnS}_4$  (11.23 eV) and  $\text{Cu}_2\text{VSnS}_4$  (11.52 eV) suggest high stability of the material after the Ti and V substitution.

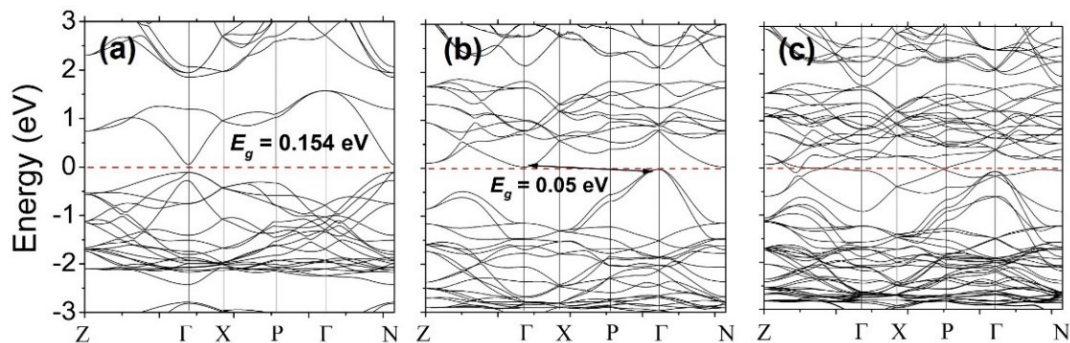


Figure 1. Band structure of (a) pristine CZTS, (b)  $\text{Cu}_2\text{TiSnS}_4$ , (c)  $\text{Cu}_2\text{VSnS}_4$

## III. CONCLUSIONS

In conclusion, Ti substitution enables to tune  $E_g$  and enhance thermodynamic stability of the material.

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